MWC(O)PHY-II/23

2023

PHYSICS

PAPER-II

Time Allowed — 3 Hours

Full Marks - 200

If the questions attempted are in excess of the prescribed number, only the questions attempted first up to the prescribed number shall be valued and the remaining ones ignored.

> Answer may be given either in English or in Bengali but all answers must be in one and same language.

Group-A

Answer any six questions.

- 1. Consider three particles each of which can be in one of three quantum states of respective energies $0 \in \mathbb{R}$ and $3 \in \mathbb{R}$. The system is at a temperature = $(k\beta)^{-1}$. Write the partition function Z if the particles obey classical MB statistics and are distinguishable. Find also Z if the particles obey B-E statistics. 5+5
- 2. Three containers, each of volume V, contain N particles of a classical, a Bose and a Fermi gas respectively at the same temperature T. State with reason which of the three containers will have the highest pressure. 10
- 3. The phase space of a one-dimensional single particle is described by its coordinate q and momentum p. Consider a rectangular phase space volume between $= q_1, q = q_2, p = p_1$ and $p = p_2$. Show that in the simple case of a free, non-interacting particle, the phase space volume remains invariant in time.

10

- 4. State the differences between depletion type and enhancement type MOSFET. Draw $I_D V_{gs}$ transfer characteristics of depletion type MOSFET. 5+5
- 5. Differentiate between n-type and p-type semiconductor. Draw and explain the transfer characteristics of a BJT in common emitter mode. 5+5
- 6. (a) What is meant by 'cross-section' of a nuclear reaction? Calculate the threshold energy of the reaction ¹⁴ $N(n, \alpha)^{11}B$.

Relevant masses are: $M(^{14}N) = 14.007550$ amu

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M(n) = 1.008983 amu

 $M(\alpha) = 4.003879 \text{ amu}$

 $M\binom{11}{B}=11\cdot012811\ \mathrm{amu}$ For guidance of WBCS Prelims , Main Exam and Interview by WBCS Gr A Officers/ 3327 oppers & Experts, WBCS Prelims and Main Mock Test (Classroom At Kolkata, Sillerise Turn Over Other Places & Online), Optional Subjects, Study Materials, Correspondence Course, etc. Call WBCSMadeEasy™ at 8274048710 / 85858543673 / 9674493673 (Sir) or mail us at mailus@wbcsmadeeasy.in. Download WBCS MADE EASY app from play store. Miscellaneous Service, Clerkship & other WBPSC Courses & Mock Test available from WBCS MADE EASY.

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- (b) Explain with reasons whether the following reactions are allowed or forbidden:
 - (i) $p \to \pi^+ + \pi^- + e^-$
 - (ii) $\pi^+ + n \to \pi^0 + k^+$

(iii)
$$p + \pi^- \to n + \pi^0$$
 (2+5)+3

- 7. (a) Determine the ground state spin parity of ${}^{19}_{9}F$ in the context of single particle shell model.
 - (b) Show that the conservation of angular momentum is not violated in β decay if the intrinsic spin of the neutrino is $\frac{\hbar}{2}$. 5+5
- 8. (a) Use uncertainty principle to calculate the radius and energy of first Bohr Orbit.
 - (b) The exciting line in an experiment is 546 nm and the stokes line is at 552 nm, find the wavelength of the anti-Stokes line.

 5+5
- 9. What is the physical significance of Fermi-level in semiconductor? Sketch the Fermi-Dirac distribution function and its derivative for T = 0 K and T > 0 K showing clearly the Fermi energy.

Group-B

Answer any seven questions.

- 10. (a) For a system of N one-dimensional harmonic oscillators, obtain the canonical partition function and calculate the average energy.
 - (b) Show that the average energy of a system in contact with a heat bath can be expressed as $\langle E \rangle = -\frac{\partial \ln z}{\partial \beta}$ where Z is the partition function and $=\frac{1}{KT}$. (5+5)+10
- 11. (a) The entropy of an ideal gas of N monoatomic molecules occupying a volume V at a temperature T is given by,

$$S(N, V, T) = Nk \log \frac{V}{N} + \frac{3}{2}Nk \left[\frac{5}{3} + \log \left(\frac{2\pi mkT}{h^2} \right) \right]$$

Two non-identical gases having N_1 and N_2 molecules, occupying volumes V_1 and V_2 respectively, are allowed to mix quasi-statically by removing the separating wall. Find the change in entropy in this mixing process. Assume that initially, the gases are at the same temperature T and have equal number densities, i.e., $\binom{N_1}{V_1} = \binom{N_2}{V_2}$.

Also assume the mass of all gas molecules to be equal.

(b) A particle in 1D has energy = $\frac{p^2}{2m} + \lambda q^4$, where q and p denote the generalized coordinate and momentum, respectively. Show that heat capacity of a gas comprising of N such particle is $C_V = \frac{3}{4}Nk$.

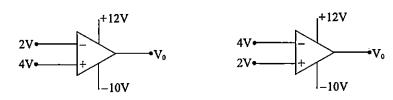
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- 12. (a) (i) In how many ways can 5 identical balls be distributed among 3 identical boxes where each box can contain any number of balls?
 - (ii) Consider a free particle inside a 1D box of length L. Calculate the number of microstates between the energy values E and + dE.
 - (b) (i) Derive Wein's displacement law from Plank's law.
 - (ii) The entropy of blackbody radiation is given by, $S = \frac{4}{3}\sigma V^{\frac{1}{4}}E^{\frac{3}{4}}$. Show that PV = $\frac{E}{3}$.
- 13. (a) What should be the input resistance, output resistance and voltage gain of an ideal OP AMP? Calculate the voltage gain of an inverting OP AMP.
 - (b) Find out the output voltage of the following two circuits:

(6+6)+8



- 14. (a) (i) Verify the Boolean identity: AC + ABC = AC
 - (ii) Verify the Boolean identity: $A + \overline{AB} = A + B$
 - (iii) Show that $\overline{AB} + \overline{A} + AB = 1$
 - (b) Implement the Boolean function $Y = AB + \overline{AC}$ using only NAND gates.
 - (c) Draw the circuit diagram of two input AND gate with the help of diodes and write the truth table of it. (3+3+3)+5+6
- 15. (a) What is basic difference between a nuclear reactor fission and a nuclear fission bomb? Explain with example the action of Moderator and control rods in a fission reactor.
 - (b) Use the semi empirical mass formula to construct the mass-parabola for isobars having same mass number A. Hence find an expression for the most stable isobar. Illustrate your answer by a suitable plot of M(A, Z) against Z. (4+3+3)+(6+2+2)
- 16. (a) Consider the one-dimensional problem of a particle in a box, i.e., a particle placed in a potential:

$$V(x) = 0$$
, for $0 < x < L$

 $=\infty$, otherwise.

Evaluate the x-p uncertainty product $(\Delta x)^2(\Delta p)^2$, for the ground state.

(b) The initial (r = 0) wave function of a free particle is described by the Gaussian wave packet

 $\psi_0 = Ae^{-\alpha x^2}$, where A and α are constants.

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- (i) Normalize the wave function $\psi_0(x)$.
- (ii) Find the wave function $\psi(x,t)$ of the free particle at a later time t.
- (iii) Calculate the probability density and hence show that the wave packet of the free particle broadens spatially with time.
- (iv) Plot the probability density at time t = 0 and time t > 0 with x.

8+(3+3+3+3)

- 17. (a) Find the degeneracy of an energy level with principal quantum number n for a hydrogen atom.
 - (b) Define group velocity. Show that group velocity of a wave packet is equal to the velocity of a particle. Using the definition of expectation value of an observable, show that $\frac{d\langle x\rangle}{dt} = \frac{\langle P_x\rangle}{m}, \text{ where the symbols have their usual meanings.} \qquad 8+(2+4+6)$
- 18. (a) What do you mean by LS- and jj-coupling? When do they occur? Irradiation of CCl₄ by 435.8 nm radiation yields Raman lines at 440.0, 441.9 and 444.7 nm. Calculate the Raman shift for line in the middle.
 - (b) What do you mean by 'population inversion'?
 - (c) Find the ratio of the rate of spontaneous and stimulated emission in terms of frequency and temperature.
 - (d) Draw the energy level diagram of ruby laser.

(2+2+4)+4+4+4

- 19. (a) Consider a particular energy band, which is filled with electrons up to the value $k = k_1$. Show that the effective number of free electrons in the band is directly proportional to the slope of the E k diagram $\left(\frac{dE}{dk}\right)_{k=k_1}$.
 - (b) Explain why the electronic specific heat associated with a free electron gas is negligibly small. What are the various contributions to the specific heat of a solid? Use Debye theory of specific heat to establish the famous Debye T^3 law. 6+(4+4+6)

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